

Effect of electronic configuration of substituents for manganese and nonstoichiometry defects on the properties of $\text{La}_{0.7}\text{Sr}_{0.3}\text{Mn}_{0.9}^{57}\text{Fe}_{0.05}\text{Me}_{0.05}\text{O}_{3+\gamma}$ (Me = Zn, Mg) manganites

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Structural and magnetic characteristics of $\text{La}_{0.7}\text{Sr}_{0.3}\text{Mn}_{0.9}^{57}\text{Fe}_{0.05}\text{Zn}_{0.05}\text{O}_{3+\gamma}$ and $\text{La}_{0.7}\text{Sr}_{0.3}\text{Mn}_{0.9}^{57}\text{Fe}_{0.05}\text{Mg}_{0.05}\text{O}_{3+\gamma}$ manganites containing Mössbauer isotope ^{57}Fe are investigated and compared. The Zn^{2+} and Mg^{2+} ions have almost identical radii (0.74 and 0.72 Å), but different configurations of electron shells ($3d^{10}$ and $2p^6$, correspondingly). Ceramic samples were sintered in air at 1473 K. They were then exposed to heat treatments at 1223 K and partial pressure of oxygen in the gas phase of $P_{\text{O}_2} = 10^{-1}$ Pa, 10^{-8} Pa, and 101.3 kPa, which ensured the production of manganites with stoichiometric oxygen content ($\gamma = 0$), with $\gamma < 0$ (containing anion vacancies) and $\gamma > 0$ (containing cation vacancies), respectively. All synthesized manganites have rhombohedral crystal structure. Mössbauer spectroscopy data correspond to Fe^{3+} ($3d^5$) ions. Non-stoichiometry index (γ) is calculated from the data on unit cell volume according to algorithm proposed earlier [1,2]. The following values of γ are obtained: $\gamma = -0.005; 0.000; 0.007; 0.008$ for Zn-containing manganites (ZnM), and $\gamma = -0.022; 0.000; 0.002; 0.005$ for Mg-containing manganites (MgM). ZnM have essentially higher values of magnetization, Curie point (T_c), and narrowest temperature interval (ΔT) of “ferromagnetic-paramagnetic” transition as compared to MgM. Their structure can be considered as more homogeneous, which corresponds to a lower value of quadrupole splitting (QS). Manganites annealed in oxygen have the narrowest ΔT , apparently due to vacancy mechanism of cation diffusion that smooths out spatial variations of the composition. Magnetization and T_c of ZnM increase in general with increasing the oxygen content, showing a plateau in the intermediate range of γ . MgM have similar dependence of T_c on γ , but their magnetization has a maximum at $\gamma = 0$ and a sharp decline at $\gamma = 0.005$, although QS at latter point is minimal. The results obtained indicate that different effect of Zn^{2+} and Mg^{2+} ions on electromagnetic characteristics of manganites is largely determined by the configuration of their electron shells.

References:

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