

Spin-orbit singlet magnetism - Induced atomic U- and Ru-moment in URu₂Si₂ and in Sr₂RuO₄

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Despite the fact that spin-orbit (s-o) interactions are well known in the atomic physics its importance in the solid-state physics was only recognized about 15 years ago with studies of *5d/4d* oxides like Sr₂IrO₄ or Sr₂RuO₄ [1,2]. Earlier scientific papers pointing out the fundamental importance of the s-o interactions also in more-discussed *3d* oxides have been rejected [3,4,5,6,7] even in the most prestigious journal as PRL or PRB with arguments that they are weak interactions, not of importance compared to *U* and *J_H* parameters of the energy size of 5 and 0.5 eV, respectively. Also, that there is no need for such interactions for explanation of known, at that time, phenomena. A "revolution" about the importance of the s-o interactions seems to start with two Phys.Rev.Lett.'s papers of Jackeli and Khaliullin in 2009 [1,2]. In the first one they interpreted INS excitations in Sr₂IrO₄ as due to s-o interactions in the Ir⁴⁺-ion in the *5d⁵* configuration [1]. In the second paper they managed to convince the PRL Editor that weak s-o interactions in the V⁴⁺(*3d¹*)-ion produce the nonmagnetic state of the whole Sr₂VO₄ oxide [2] in contrary to my earlier submissions from 1997-2010. This nonmagnetic state is clearly shown in Fig. 1 of Ref. [4] from 1999 with detailed examples of the V⁴⁺ ions in BaVS₃ [6,7] and in Sr₂VO₄.

In this contribution I will discuss *3d/4d/5d* compounds/oxides underlying i) formation of the charge ionic state with the well-defined integer valency, like U⁴⁺ and Ru⁴⁺ ions, ii) the discrete quasi-atomic crystal-field+s-o spin-orbital low-energy electronic structure and iii) the preservation of this low-energy (below 1 meV) quasi-atomic *3d/4d/5d* electronic structure also in solid crystals. As the s-o effects I will discuss the formation of the singlet nonmagnetic ground state of the U⁴⁺(*5f²*) and Ru⁴⁺(*4d⁴*) ions and their preservation in URu₂Si₂ [3] and in Sr₂RuO₄. In the presented approach, which one could call as Quantum Atomistic Solid-State Theory (QUASST), the orbital moment, as the s-o effect, will be discussed in NiO (Ni²⁺) [5] and in FeBr₂ (Fe²⁺) [8].

References:

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