

## Electronic structure of $R\text{Ag}_2\text{Ge}_2$ ( $R = \text{Pr}, \text{Nd}$ ) compounds

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The electronic structure of ternary  $R\text{Ag}_2\text{Ge}_2$  ( $R = \text{Pr}, \text{Nd}$ ) compounds, which crystallize in the tetragonal  $\text{ThCr}_2\text{Si}_2$  - type structure, were studied by X-ray photoemission spectroscopy. The magnetic data indicate that  $\text{PrAg}_2\text{Ge}_2$  remains paramagnetic down to 1.9 K, whereas  $\text{NdAg}_2\text{Ge}_2$  orders antiferromagnetically at about 2 K. The XPS results clearly show that the valence bands in both germanides consist mainly of the  $\text{Ag } 4d$  band. The XPS spectra of  $\text{Pr}$  and  $\text{Nd}$   $3d_{5/2}$  and  $3d_{3/2}$  core levels were analyzed in the framework of the Gunnarsson-Schönhammer model [1] in order to derive information on the hybridization of  $4f$  orbital with the conduction band. Separation of the XPS peaks, based on the Doniach-Šunjić theory [2], yielded the ratio  $r = I(f^{n+1})/[I(f^n) + I(f^{n+1})]$  being equal to 0.09 for  $\text{PrAg}_2\text{Ge}_2$  and 0.17 for  $\text{NdAg}_2\text{Ge}_2$ , which corresponds to the hybridization energy of 58.1 and 84.4 eV, respectively. The obtained values indicate a stability of the  $4f$  shell in these two compounds, in good agreement with the magnetic data.

[1] O. Gunnarsson, K. Schönhammer, Phys. Rev. B, **28** (1983) 4315.

[2] S. Doniach, M. Šunjić, J. Phys. C, **3** (1970) 285.

13.4 cm

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9.7 cm